

Photolysis of Indomethacin in Methanol

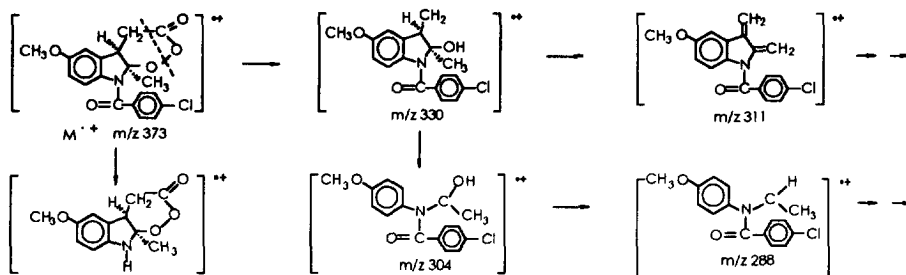
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Abstract: A novel photo-oxidation product containing a six-membered 1,2-dioxane ring was isolated from the photolysis of indomethacin in methanol and a possible mechanism is proposed. © 1997, Published by Elsevier Science Ltd. All rights reserved.

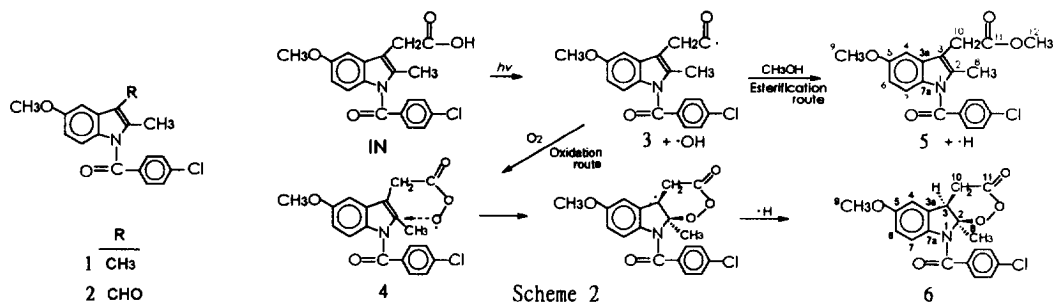
Indomethacin (IN) is used as an anti-inflammatory, antipyretic and analgesic therapeutic agent, however, it is photochemically labile. Weedon and Wong found that irradiation of IN in benzene using a medium pressure mercury lamp yielded **1** by decarboxylation of the acetic acid side chain; in methanol yielded **2** by further oxidation¹. Similar results were observed by Dabestani, et al.²

We now report that when IN in methanol (27.9 m mol) was irradiated by daylight for 7 d, two major photolytic products, **5** and **6** were isolated by preparative HPLC³ and their structures characterized^{4,5}. The EI-MS of **6** revealed some interesting characteristics (Scheme 1). ¹H NMR of **6** shows the signal of methine hydrogen attached to C-3 at 2.54 ppm and methyl attached to C-2 at 1.74 ppm which are in accord with a *syn* relationship⁶. ¹³C NMR of **6**, C-2 appears at 83.4 ppm and C-3 at 41.3 ppm which strongly imply both carbons becoming saturated. Attribution of protons of 1,2-dioxan-3-one ring system is achieved through examination of HETCOR and DEPT ¹H-¹³C connectivities. These results clearly indicate that **6** contains an indoline ring⁷.



Scheme 1

Irradiation of IN in N_2 -gassed or air-saturated benzene, several radical species were detected². Thus the photolysis of IN in organic solvents via radical pathways were assumed. In our study, the formation of **5** and **6** was apparently not gone through the decarboxylation process; instead, we postulate that they were formed by acyl cleavage of the carboxylic group by sharing with a common intermediate, an acyl radical, **3** (Scheme 2). Starting from **3**, **5** was produced via an esterification route; **6** via an oxidation route by reacting with a molecular oxygen. The formation of **6**, 9-(4-chlorobenzoyl)-6-methoxy-9a-methyl-4,4a-dihydro-1,2-dioxo-9-aza-fluoren-3-one, represents one of an unprecedented example of intramolecular peracetate radical **4** addition to the 2 position of the indole ring leading to a six-membered 1,2-dioxane derivative.



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Reference and notes:

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- HPLC: Hitachi L-6200 pump, L-4200 detector at 254nm, Nucleosil 7 C₁₈ column (250*4.6mm i.d.), mobile phase, CH₃OH:H₂O (1% acetic acid) = 75:25, flow rate: 2ml/min. Retention time: **6**, 7.50 min; **5**, 15.94 min. Yield of **6**=11.2 %; **5**=20.9 % (by integration) approaching maximum after 30 d of irradiation.
- Methyl 1-(4-chlorobenzoyl)-5-methoxy-2-methylindolyl-3-acetate, **5**: Pale yellow needles, mp=85-87 °C. ¹H NMR δ (ppm) CDCl₃ (500Hz): 6.94 (d, ¹H, J=8.7Hz, H-4), 6.62-6.65 (dd, 1H, J=8.8Hz, 2.6Hz, H-6), 6.83-6.86 (d, 1H, J=8.9Hz, H-7), 2.35(s, 3H, H-8), 3.80 (s, 3H, H-9), 3.64 (s, 2H, H-10), 3.68 (s, 3H, H-12). ¹³C NMR δ (ppm) CDCl₃ (500MHz): 130.53(C-2), 112.39 (C-3), 130.69 (C-3a), 101.22 (C-4), 155.95 (C-5), 111.48 (C-6), 114.83 (C-7), 139.08 (C-7a), 13.19 (C-8), 55.57 (C-9), 29.98 (C-10), 171.18 (C-11), 52.00 (C-12). IR (KBr): 1736cm⁻¹ (s, C=O). MS (70ev, m/z (%)): 371 (M+, 75), 327 (6), 312 (17), 139 (100).
- 9-(4-Chlorobenzoyl)-6-methoxy-9a-methyl-4,4a-dihydro-1,2-dioxo-9-aza-fluoren-3-one, **6**: Yellow crystals mp=217-218 °C. ¹H NMR δ (ppm) CD₃COCD₃ (500MHz): 7.17 (d, 1H, H-4), 6.78(dd, 1H, J=8.8, 2.6 Hz, H-6), 6.84 (d, 1H, J=8.7 Hz, H-7), 2.54 (d, 1H, H-3), 1.74 (s, 3H, H-8), 3.79 (s, 3H, H-9), 3.19(d, 2H, J=6.1 Hz, H-10). ¹³C NMR δ (ppm) CD₃COCD₃ (500 MHz): 83.44 (C-2), 41.28 (C-3), 136.21 (C-3a), 110.55 (C-4), 158.05 (C-5), 117.01 (C-6), 118.51 (C-7), 137.86 (C-7a), 19.16 (C-8), 56.05 (C-9), 40.48 (C-10), 171.65 (C-11). IR (KBr): 1796 cm⁻¹ (s, C=O). MS(70 ev, m/z (%)): 373 (M+, 100), 330 (13), 311(2), 288 (2), 234 (31), 217 (5), 190 (22), 175 (3), 139 (37).
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